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Write your answer in the space provided. Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that, perhaps, the grass is dying.

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In your notebook, answer the following SECTION 7.1 IONS For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion. a. Ba 2. How many valence electrons does each of the following atoms have? a. gallium b. fluorine c. selenium 3.

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Science - Chemistry Core

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Answer the following questions in the space provided, 19. 20. 21. 10 Describe two ways biotechnology can help treat diseases. insaliO Name three factors that decrease crop productivity. Explain how scientists know that water once existed on the surface on Mars. Core Teaching Resources

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experiments; Core Teaching Resource: 45; Small Scale Lab Manual: p.4-5, All 41 experiments C-2.1 Illustrate electron configurations by using orbital notation for representative elements. SE/TE: 133-135, 164, 166, 442 TE: Class Activity: 134 TR: Core Teaching Resource: 117 TECH: Interactive Textbook: Simulation 2 C-2.2 Summarize atomic properties

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Name Date 11. The number of sublevels in an energy level is equal to the square of the principal quantum number of that energy level. 12. The maximum number of electrons that can occupy the fourth principal energy level of an atom is 32.

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