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The energy required to break a bond, and the energy released when a bond is formed. An insulated container used to study reactions at a constant volume.  $\delta G = \delta H - T\delta S$ , where  $\delta H$  = change in enthalpy and  $\delta S$  = change in entropy. The point at which the vapor pressure of a liquid equals 1 atm.

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representatively, it is imperative that you be well versed in the general principles of E1, E2, Sn1, Sn2 reactions and of carbonyl compounds. Nomenclature will be covered. Spectroscopy will be covered, but lightly (only about 3 problems, one H-NMR, one C-NMR, one IR.) By my count there are about 45 problems that I would consider very straightforward,

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Page 4 Property of ACS USNCO - Not for use as USNCO Local Section Exam after March 31, 2018 13. What state of matter corresponds to the diagram below? (A) Gas (B) Liquid (C) Amorphous solid aluminum block? (The specific heat capacity of Al is (D) Crystalline solid 0.900 J g

### **2018 U.S. NATIONAL CHEMISTRY OLYMPIAD**

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National Norms The data included here are abbreviated norms, presenting only percentile rank as a function of raw score and overall test statistics. Complete norms are computed, printed, and subsequently shipped with orders for examinations as soon as sufficient data is received at the Examinations Institute.

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