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Terms in this set (11) saturated. solution containing all the dissolved solute possible at given conditions of temperature and pressure. unsaturated. solution containing less dissolved solute than the maximum amount that can be dissolved at given conditions of temperature and pressure. supersaturated.

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Chemistry Note Taking Guide: Episode 1002. Terms in this set (6) Saturated. Solution containing all the dissolved solute possible at given conditions of

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temperature and pressure. Unsaturated. Solution containing less dissolved solute than is normally possible at given conditions of temperature and pressure.

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Note Taking Guide Episode 1003. Terms in this set (9) Molarity (M) expresses solution concentration. Molarity formula.  $M = \text{moles of solute} / \text{liters of solution}$ . Colligative properties. any of the properties of a solution that change when the concentration of the solution changes (depend on the # of particles

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dissolved in a given mass of solvent)

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tells the relative number of atoms of each element in a compou.... ( covalent bonding) formula represents the number of atoms of.... (ionic bonding) formula represents the simplest ratios of pos.... the attraction that binds oppositely charged ions together.

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Students learn how to distinguish  
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